Russian Journal of Organic Chemistry, Vol. 38, No. 10, 2002, pp. 1540–1541. Translated from Zhurnal Organicheskoi Khimii, Vol. 38, No. 10, 2002, pp. 1591–1592.

Original Russian Text Copyright © 2002 by Bol'shakov, Klimova, Rudaya, Yurre, Shamanin, Skorokhodov.

SHORT COMMUNICATIONS

## Mesogenic Triad with a Benzoyl Group\*

M. N. Bol'shakov<sup>2</sup>, N. V. Klimova<sup>1</sup>, L. I. Rudaya<sup>1</sup>, T. A. Yurre<sup>1</sup>, V. V. Shamanin<sup>2</sup>, and S. S. Skorokhodov<sup>2</sup>

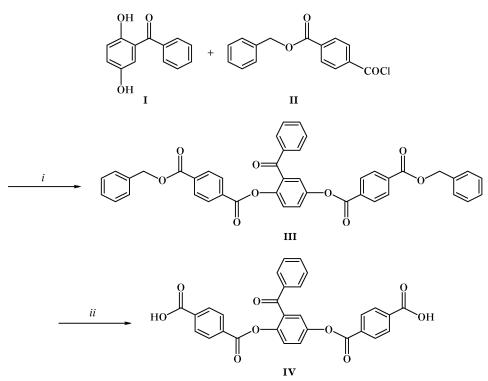
<sup>1</sup> St. Petersburg State Institute of Technology, Moskovskii pr. 26, St. Petersburg, 198013 Russia

<sup>2</sup> Institute of High-Molecular Compounds, Russian Academy of Sciences, St. Petersburg, 199004 Russia

Received January 21, 2002

The reaction of 2,5-dihydroxybenzophenone (I) with benzyl 4-chloroformylbenzoate (II) in aqueous– organic medium in the presence of a phase-transfer catalyst leads to formation of 2,5-bis(4-benzyloxycarbonylbenzoyloxy)benzophenone (III). Debenzylation of ester III [1] gives 2,5-bis(4-carboxybenzoyloxy)benzophenone (IV) (Scheme 1). Compound IV is a rigid mesogenic triad possessing a photo- and chemically active moiety, a carbonyl group. Compound I was synthesized by Friedel–Crafts benzoyla-

Scheme 1.



i: Bu<sub>4</sub>NBr or BuMe<sub>3</sub>NCl, 4% NaOH, H<sub>2</sub>O-CH<sub>2</sub>Cl<sub>2</sub>; ii: HBr (AcOH, CF<sub>3</sub>COOH) [1].

This study was financially supported by the Russian Foundation for Basic Research (project no. 01-03-32292) and by the Program "Higher School Research in the Priority Fields of Science and Technics" (project no. 203.02.06.015).

tion of 1,4-dimethoxybenzene, followed by successive dealkylation first with  $AlCl_3$  to 2-hydroxy-5-methoxybenzophenone and then with HI in  $Ac_2O$  [2]. Benzyl 4-chloroformylbenzoate was prepared by the procedure described in [3].

2,5-Bis(4-benzyloxycarbonylbenzoyloxy)benzophenone (III). A solution of 5.60 g (20.4 mmol) of compound II in 60 ml of  $CH_2Cl_2$  was added dropwise to a solution of 2.75 g (12.8 mmol) of 2,5-dihydroxybenzophenone in 60 ml of 4% aqueous sodium hydroxide containing 0.28 g of tetrabutylammonium bromide. The mixture was stirred for 30 min at room temperature in an inert atmosphere. The organic phase was separated, the solvent was removed, and the residue (6.3 g of compound **III**) was twice recrystallized from isopropyl alcohol. Yield 3.14 g (45%), mp 118-120°C. R<sub>f</sub> 0.79 (Silufol UV-254, solvent CH<sub>2</sub>Cl<sub>2</sub>, eluent alcohol-benzene, 1:10). IR spectrum (KBr), v, cm<sup>-1</sup>: 2920 (CH<sub>2</sub>), 1755 (C=O, ester), 1680 (C=O, ketone). Found, %: C 74.17; H 4.07. C<sub>43</sub>H<sub>30</sub>O<sub>9</sub>. Calculated, %: C 74.77; H 4.38.

**2,5-Bis(4-carboxybenzoyloxy)benzophenone** (**IV**). Compound **III**, 2.82 g (4.08 mmol), was dissolved in 39 ml of CF<sub>3</sub>COOH, and 3.1 ml of a 36% solution of HBr in CH<sub>3</sub>COOH was added dropwise. The mixture was stirred for 24 h at room temperature, and the precipitate (1.76 g) of compound **IV** was filtered off. An additional portion of the product, 0.22 g, was isolated from the filtrate by precipitation with acetone. The two portions were combined and washed with acetone. Yield 1.96 g (94%), mp 305–307°C.  $R_f$  0.45 (Silufol UV-254, solvent DMF, eluent benzene–ethyl acetate, 1:1). IR spectrum (KBr), v, cm<sup>-1</sup>: 1755 (C=O, ester), 1720 (C=O, acid), 1680 (C=O, ketone). <sup>13</sup>C NMR spectrum (DMF- $d_7$ ),  $\delta_C$ , ppm: 194.2 (C=O); 167.9, 167.6 (COOH); 165.3, 165.0 (OCO).

The IR spectra were recorded on a Bruker IFS-88 spectrometer. The <sup>13</sup>C NMR spectrum was obtained on a Bruker AM-500 instrument at 125 MHz. Elemental analysis was performed on a Perkin–Elmer automatic analyzer.

## REFERENCES

- Galli, G., Chiellini, E., Ober, Ch.K., and Lenz, R.W., Makromol. Chem., 1982, vol. 183, no. 11, pp. 2693– 2708.
- Percec, V., Bae, J.-Y., Zhao, M., and Hill, D.H., J. Org. Chem., 1995, vol. 60, pp. 1066–1069.
- Hasslin, H.-W., Droscher, M., and Wegner, G., Makromol. Chem., 1980, vol. 181, no. 2, pp. 301–323.